

National Institute of Standards & Technology

Certificate of Analysis

Standard Reference Material® 913a

Uric Acid

This Standard Reference Material (SRM) is certified as a chemical of known purity. It is intended primarily for use in the calibration and standardization of procedures for uric acid determinations employed in clinical analysis and for routine critical evaluation of the daily working standards used in these procedures. A unit of SRM 913a consists of one bottle containing 10 g of crystalline uric acid.

Certified Value and Uncertainty: The certified value is based upon the results from several analytical techniques designed to measure impurities and on scientific judgement of these results. The certified chemical purity presented in Table 1 was determined by measuring the mass fractions of impurities, including water and residue from ashing, summing the impurities, adding an allowance for undetected impurities, and subtracting this sum from 100 %. A NIST certified value is a value for which NIST has the highest confidence in its accuracy in that all known or suspected sources of bias have been investigated or accounted for by NIST.

Table 1. Certified Chemical Purity and Uncertainty for Uric Acid (Mass Fraction)

Purity: 99.6 % ± 0.1 %

The results are expressed as the certified value \pm the expanded uncertainty. The uncertainty in the certified value is equal to $U = ku_c$, where, u_c , is the combined standard uncertainty calculated according to the ISO Guide [1] and, k = 1.97, is the coverage factor. The expanded uncertainty is intended to represent a 95 % confidence interval and reflects the combined effects of measurement imprecision, variability in concentrations between vials, and any systematic differences between techniques when more than one method has been used [1].

Reference Values: Reference values are noncertified values that are the best estimates of the true values; however, the values do not meet NIST criteria for certification. Such values are provided with associated uncertainties that may reflect only measurement precision, may not include all sources of uncertainty, or may reflect a lack of sufficient statistical agreement among multiple analytical methods. Reference values for mass loss upon drying, residue after ashing, and molar extinction coefficients are provided in Table 2.

Expiration of Certification: The certified value is valid until **01 January 2010** within the measurement uncertainties specified, provided the SRM is handled and stored in accordance with the instructions given in this certificate. However, the certification is nullified if the SRM is contaminated or modified.

Maintenance of SRM Certification: NIST will monitor this SRM over the period of its certification. If substantive technical changes occur that affect the certification before the expiration of this certificate, NIST will notify the purchaser. Return of the attached registration card will facilitate notification.

Analytical measurements at NIST were performed by S.S-C. Tai, L.T. Sniegoski, and M.J. Welch of the NIST Analytical Chemistry Division.

The support aspects involved in the preparation, certification, and issuance of this SRM were coordinated through the NIST Standard Reference Materials Program by J.C. Colbert.

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Statistical analysis of data was provided by N-F. Zhang of the NIST Statistical Engineering Division.

The overall direction and coordination of the technical activities were under the chairmanship of M.J. Welch of the NIST Analytical Chemistry Division.

NOTICE AND WARNING TO USER

Storage: SRM 913a should be stored in its original bottle at temperatures between 20 °C and 25 °C. It must be tightly re-capped after use and protected from excessive moisture and light.

Instructions for Use: SRM 913a is not hygroscopic under ordinary conditions of storage as described above, and can be used without preliminary drying.

SOURCE AND ANALYSIS

Source of Material: The SRM material was obtained from a commercial supplier.

Analytical Approach: All analyses for the certified and information values were performed at NIST. Because uric acid does not melt, differential scanning calorimetry could not be used as an overall measure of the purity. Instead, impurities were tested for using a variety of techniques that included: liquid chromatography (LC), direct probe mass spectrometry (DP-MS), mass loss upon drying, ashing at 800 °C, and ultraviolet (UV) spectroscopy. No significant impurities were found by the techniques LC, DP-MS, or UV spectroscopy. The results for the drying and ashing analyses are shown in Table 2. Other low level impurities may be present but were not detected by the techniques used here. The certified purity is determined by subtracting from 100 %, the sum of the mass loss upon drying, the residue after ashing, and an allowance of 0.1 % for undetected impurities.

Molar extinction coefficients were measured for solutions of SRM 913a in a glycine-NaOH aqueous buffer (pH 9.6) and 0.0017 mol/L ammonia (pH 10.1) at 234 nm and 292 nm. The results for the molar extinction coefficients are also shown in Table 2.

Table 2. Reference Values for Selected Properties of SRM 913a

Components of Impurity	1,140	Mass Fraction (in %)	
Mass loss upon drying Residue after ashing	0.17 0.09	± 0.02 ± 0.01	
UV Absorbance of Uric Acid	Molar Extinction	Molar Extinction Coefficients	
	234 nm	292 nm	
In glycine-NaOH buffer, pH 9.6	9.789 ± 149	$12\ 612\pm 80$	
In ammonia solution, pH 10.1	$9\ 847 \pm 105$	$12\ 756\pm93$	

REFERENCE

[1] Guide to the Expression of Uncertainty in Measurement, ISBN 92-67-10188-9, 1st Ed. ISO, Geneva, Switzerland, (1993); see also Taylor, B.N. and Kuyatt, C.E., "Guidelines for Evaluating and Expressing the Uncertainty of NIST Measurement Results," NIST Technical Note 1297, U.S. Government Printing Office, Washington DC, (1994); available at http://physics.nist.gov/Pubs/.

Users of this SRM should ensure that the certificate in their possession is current. This can be accomplished by contacting the SRM Program at: telephone (301) 975-6776; fax (301) 926-4751; e-mail srminfo@nist.gov,

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or via the Internet http://www.nist.gov/srm.

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